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Key indicators

Single-crystal X-ray study T = 298 K Mean σ (C–C) = 0.005 Å R factor = 0.059 wR factor = 0.158 Data-to-parameter ratio = 12.2

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1-(4-Methylphenylsulfonyl)-5-nitro-1*H*benzimidazole

In the title compound, $C_{17}H_{15}N_3O_4S$, the mean plane of the benzimidazole fragment and the benzene ring of the *p*-tosyl group make a dihedral angle of 86.72 (14)°. The molecules are linked by weak C-H···O intermolecular hydrogen bonds into chains parallel to the *b* axis. The crystal packing is further stabilized by π - π stacking interactions.

Comment

The title compound, (I), is analogous to 1-(4-methylphenyl-sulfonyl)5-nitro-2-[(E)-prop-1-enyl]-1H-benzimidazole, (II) (Rashid *et al.*, 2006).



In (I) (Fig. 1), all bond lengths and angles are in normal ranges (Allen *et al.*, 1987) and comparable to those observed in (II). The mean plane of the benzimidazole fragment and the benzene ring of the *p*-tosyl group make a dihedral angle of 86.72 (14)°. The crystal packing (Fig. 2) exhibits weak intermolecular C–H···O hydrogen bonds (Table 2), which link the molecules into chains parallel to the *b* axis, and $\pi \cdot \cdot \pi$ stacking interactions with a shortest distance of 3.645 (2) Å between the centroids of the N1/N2/C1/C6/C7 and (C1–C6)ⁱⁱ rings [symmetry code: (ii) 1 - x, -y, -z].

Experimental

Equimolar quantities of 4-nitro-1,2-phenylenediamine and methanoic acid were refluxed in 4N HCl to synthesize 5-nitrobenzimidazole. The title compound was then synthesized by a known method (Hasan *et al.*, 1992) and recrystallized from ethanol (yield 38%, m.p. 446 K).

Crystal data $C_{14}H_{11}N_3O_4S$ $M_r = 317.32$ Monoclinic, $P2_1/n$ a = 10.630 (3) Å b = 9.950 (3) Å c = 13.662 (4) Å $\beta = 104.577$ (4)° V = 1398.5 (7) Å³

Z = 4 $D_x = 1.507 \text{ Mg m}^{-3}$ Mo K α radiation $\mu = 0.25 \text{ mm}^{-1}$ T = 298 (2) K Block, colourless $0.29 \times 0.24 \times 0.19 \text{ mm}$ Received 21 November 2006 Accepted 11 December 2006

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Figure 1

The molecular structure of (I), with displacement ellipsoids drawn at the 50% probablity level.

Data collection

Bruker SMART APEX CCD area-6561 measured reflections detector diffractometer 2453 independent reflections ω scans 1871 reflections with $I > 2\sigma(I)$ Absorption correction: multi-scan $R_{\rm int}=0.029$ (SADABS; Bruker, 2000) $\theta_{\rm max} = 25.0^{\circ}$ $T_{\rm min} = 0.929, \ T_{\rm max} = 0.953$

Refinement

$w = 1/[\sigma^2(F_o^2) + (0.0583P)^2]$
+ 1.2495P]
where $P = (F_0^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.19 \text{ e } \text{\AA}^{-3}$
$\Delta \rho_{\rm min} = -0.32 \text{ e } \text{\AA}^{-3}$
Extinction correction: SHELXL97
Extinction coefficient: 0.0012 (6)

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C7-H7A\cdotsO1^{i}$	0.93	2.31	3.239 (5)	173
6	1.1			

Symmetry code: (i) x, y + 1, z.

All H atoms were postioned geometrically [C-H 0.93-0.96 Å] and constrained to ride on their parent atoms with $U_{iso}(H) = 1.2$ - $1.5U_{eq}(C).$





Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.

Bruker (2000). SADABS (Version 2.01), SMART (Version 5.630) and SAINT (Version 6.36a). Bruker AXS Inc., Madison, Wisconsin, USA.

Hasan, M., Rashid, N., Malik, F., Akhtar, K., Osman, S. & Duddeck, H. (1992). J. Chem. Soc. Pak. 14, 54-59.

Nardelli, M. (1995). J. Appl. Cryst. 28, 659.

Rashid, N., Hasan, M., Yusof, N. M. & Yamin, B. M. (2006). Acta Cryst. E62, 05455-05456.

Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97, University of Göttingen, Germany.

Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Bruker AXS, Inc., Madison, Wisconsin, USA.

Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.